

Metals with several oxidation states:

<u>Element</u>	<u>Ion Formula</u>	<u>Systematic Name</u>	<u>Common Name</u>
Chromium	Cr^{+2}	chromium (II)	
	Cr^{+3}	chromium (III)	
	Cr^{+6}	chromium (VI)	
Manganese	Mn^{+2}	Manganese (II)	
	Mn^{+3}	Manganese (III)	
	Mn^{+4}	Manganese (IV)	
	Mn^{+6}	Manganese (VI)	
	Mn^{+7}	Manganese (VII)	
Iron	Fe^{+2}	Iron (II)	ferrous
	Fe^{+3}	Iron (III)	ferric
Cobalt	Co^{+2}	Cobalt (II)	
	Co^{+3}	Cobalt (III)	
Nickel	Ni^{+2}	Nickel (II)	
	Ni^{+3}	Nickel (III)	
Copper	Cu^{+1}	Copper (I)	cuprous
	Cu^{+2}	Copper (II)	cupric
Gold	Au^{+1}	Gold (I)	aurous
	Au^{+3}	Gold (III)	auric
Mercury	Hg^{+1}	Mercury (I)	
	Hg^{+2}	Mercury (II)	
Tin	Sn^{+2}	Tin (II)	stannous
	Sn^{+4}	Tin (IV)	stannic
Lead	Pb^{+2}	Lead (II)	plumbous
	Pb^{+4}	Lead (IV)	plumbic